# HOME/SCHOOL CONNECTION

### INVESTIGATION 3: FIZZ QUIZ

Baking soda (sodium bicarbonate—NaHCO $_3$ ) reacts with acid. One of the products is carbon dioxide (CO $_2$ ). You can use a baking-soda solution to test unknown liquids to see if they are acids. If CO $_2$  bubbles form when you mix the two solutions, the unknown probably contains an acid.

#### Materials

- Baking soda
- 1 Tablespoon
- Measuring cup
- Water
- Unknown liquids
- Small glass for testing
- 1 Spoon

#### Directions

- Put 1 heaping tablespoon of baking soda in the measuring cup.
- Add water to the 1-cup level. Stir to dissolve the soda.
- 3. Put a small amount of the baking-soda solution in a glass.
- 4. Add an equal amount of unknown solution. Record your observations.

## Things to try

Fruit juices (particularly citrus)

Vinegar

Vitamin C dissolved in water

Coffee

Soda drinks

Unknown	Observations